### **Carson Valley and Arsenic**

### Distribution of Arsenic in Southern Carson Valley: What We Currently Know





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Carson River Watershed Management Forum March 10, 2020



# PROBLEM

- Water supply wells in northern Carson Valley were decommissioned due to elevated arsenic concentrations (Carl Ruschmeyer, January 2013, Douglas County Public Works Director, verbal communication).
- Town of Minden is currently providing water from their production wells to supply water lost to consumers by the decommissioning of the wells in northern Carson Valley
- Ultimate Question: Will increasing pumping rates from water supply wells in Minden mobilize arsenic toward them thereby degrading water quality?



## **Conceptual Model: Arsenic in Groundwater**



Arsenic is naturally occurring. Enrichment can occur due to favorable hydrogeology, geochemical conditions, and long flow paths

Deep wells can tap groundwater that can contain the reduced form of inorganic As(III) – the more mobile form



Water Years 1960 – 2015

Distribution of Arsenic Concentrations in Carson Valley

### Filtered and Unfiltered





Water Years 2006 – 2015

## Distribution of Arsenic Concentrations in Carson Valley

### Filtered and Unfiltered





### **Status and Sampling Effort - 2019**

- Limited arsenic concentration data exist over the past 10-20 years for areas upgradient and surrounding production wells: Update arsenic concentration distribution assessment within the target study area (n = 9 wells)
- Redox conditions may be mixed and above what might be reducing for arsenic but speciation data are unavailable: Analyzed each sample collected from domestic wells for redox chemistry (DO, Nspecies, Mn, Fe, As-species, SO<sub>4</sub>-H<sub>2</sub>S), DOC, and chloride
- Collect samples during minimal pumping pressure (March/April 2019)



# Minden Well #8

Collaboration with water purveyors (FY2018) -

Arsenic concentrations above 10 µg/L (MCL) before purge; after purge, concentrations fall below MCL (Jeff Cady, Town of Minden, Water Operations Manager)

Arsenic in-situ remediation study (2010) with speciation data just north of Well #8

Well #8 Sampled for a suite of analytes in 2013 as part of the USGS NAWQA Program



110 – 170 250 – 270 310 – 330 <u>43</u>0 - 510

Screened intervals (below land surface)

USGS 2013 (source water) Arsenic = 9.0 µg/L



## **Domestic Well Selection Process**

Aquifer Penetration Depth Proximity to municipal supply well Location within general groundwater flow path Permission by well owner to sample well









Sample Well





## **Depth Within Aquifer Representation**



# Samples Collected

9 Environmental Samples
1 Field Blank (no speciation)
3 Sequential Replicates (all parameters)
Arsenic Oxidizing Bacteria (Farah Ansari, Pakistan/CSU)



Jena and Jonathan processing samples in domestic pump house.

Jonathan preparing to sample for arsenic speciation. Septum used to preserve redox conditions during sampling.





Faral

# Analyses

Field Parameters (DO, pH, SC, temp)

Nitrogen Species (NO<sub>3</sub>/NO<sub>2</sub>/NH<sub>3</sub>)

Manganese and Iron

Arsenic Speciation (AsV/AsIII)

Sulfate (+H<sub>2</sub>S sniff test)

Chloride, Phosphate, and DOC



**Rinsing the sampling line** 

Samples were analyzed by USGS NWQL, USGS Trace Element Lab, and Brooks Applied Labs



#### During the 2019 sampling effort, higher As along eastern edge of S. Carson Valley





Water Years 2006 – 2019

Arsenic Concentrations

Filtered and Unfiltered

Municipality and USGS As Remediation Study and 2019 Data





#### Higher arsenic concentrations observed along the east side of the valley (east of East Valley Road)

<5 ug/L As

>10 ug/L As



Provisional data, subject to revision

As (III)

#### Arsenic Concentration with Depth – Phase I & II



#### Arsenic Concentration with Depth – with 2019 Data



Arsenic Concentration, micrograms per liter





### Arsenic and Iron (Fe)

Arsenic is attracted to iron oxide. Iron oxide is the rust colored material observed as a coating in basin-fill aquifer material (picture below). Under conditions oxidizing to iron, the arsenic will not be as mobile; under conditions that do not support iron rust, arsenic can have greater mobility in the groundwater (Thiros and others, 2014, USGS Circular 1358).



Photo Credit: Core of basin-fill sediment, Susan Thiros, USGS, 2014.

*Note*: Relation in top figure is dominated by As(III) data point; reductive dissolution of metal oxides is likely liberating As into groundwater. Relation in bottom figure is not significant (p>0.05) but suggests that iron may be undergoing some degree of dissolution and releasing As(V) into groundwater (further data needed to confirm hypothesis).



# Study Findings

- Higher arsenic concentrations appear to be occurring along east side of valley; upgradient from many public supply wells
- All samples collected were analyzed for parameters known to be important to the mobility of arsenic

Iron may be influencing the mobility of arsenic in some areas/portions of the aquifer

- Arsenic primarily observed as As(V); except in the deepest well, screened below 500 ft below land surface, which was found to support As(III)
  - Importance: Generally, As(III) has greater mobility than As(V)

